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The lower solubility in water of $\text{Fe}(\text{OH})_3$ ($K_{sp} = 6 \times 10^{-38}$), $\text{Fe}(\text{OH})_2$ ($K_{sp} = 2 \times 10^{-15}$) and FePO_4 ($K_{sp} = 1 \times 10^{-22}$) compared to $\text{Fe}_2(\text{SO}_4)_3$ ($K_{sp} = 1 \times 10^{-5}$) and the content of phosphorus and sulfur in the backwashing material suggest that iron hydroxide and

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phosphate were the main iron precipitates in the RPPB during the study.

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